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3. (a) law (states a consistently observed phenomenon, can be

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used for prediction); (b) theory (a widely accepted explanation of the behavior of matter); (c) hypothesis (a tentative explanation, can be investigated by experimentation) 5. <https://studyres.com/doc/2831088/openstax---chemistry---answer-key> read more.

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CHAPTER 3 ANSWERS AND EXPLANATIONS. Multiple-Choice. 1. D This element has five peaks, meaning a total of five subshells. The final peak, which would be located in the 3p subshell, is slightly higher than the 3s peak to the left of it. A full 3s peak has two electrons, therefore there must be at least three electrons in the 3p subshell. The element that best fits this is phosphorus.



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## **CHAPTER 3 ANSWERS AND EXPLANATIONS - Atoms, Elements, and ...**

CHAPTER 3 TEST Class Atoms: The Building Blocks of Matter

MULTIPL CHOICE On the line at the left of each statement, write the letter of the choice best completes the statement or answers the question. The behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

### **San Ramon Valley High School**

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