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Chapter 3 Scientific Measurement Packet

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Chapter 3 – Scientific Measurement.
Jennie L. Borders. Section 3.1 –
Measurements and Their Uncertainty. A
measurement is a quantity that has both
a number and a unit. The unit typically
used in the sciences are those of the
International System of Measurements
(SI). In scientific notation, a given
number is written as the product of two

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numbers: a coefficient and 10 raised to a power.

Chapter 3 - Scientific Measurement

Chapter 3 - Measurement and

Uncertainty 3.1 Note Packet-

Measurement and Uncertainty 2.

INSTRUCT: Discuss Significant Figures in Measurements (10 minutes) Targeted

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Resources Point out that the concept of significant figures applies only to measured quantities. If students ask why an estimated digit is

Ch. 3 Completed Section 3.1 Measurements and Their ...

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Scientific Measurement Worksheets & Teaching Resources | TpT

Judy_Walley. Chemistry Chapter 3: Scientific Measurement. Measurement. scientific notation. accuracy. precision. a quantity that has both a number and unit. an expression of numbers in the form $m \times 10^n$ where m is equal.... a

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measure of how close a measurement comes to the actual or tr....

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Scientific Measurement Quantifying Matter 3.1 using and expressing Measurements essential Understanding In science, measurements must be

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accurate, precise, and written to the correct number of significant figures. reading Strategy Venn Diagram A Venn diagram is a useful tool in visually organizing related

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 C Measure of how close a series of measurements are to one another a.

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accuracy 2. A Measure of how close a measurement comes to the actual value b. measurement 3. F Digits in a measurement that are known plus one that is estimated c. precision 4. E A value determined in the laboratory d. scientific notation 5.

Test Review key - Chapter 3 Test

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Review Homework Name Hour ...

Chapter 3 Notes, Slides 20-24: Scientific Notation (**only moving places, no calculations) Worksheet: Scientific Notation & Operations (text version, back of sig figs worksheet) Chapter 3 Notes, Slides 25-28: SI Units Video: Powers of 10 Chapter 3 Notes, Slide 29: Start converting! Worksheet: Units of

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Measurement Table. Day 10 - IPOD #3 ...

McLaughlin, Kimberly / Matter & Measurement

CHAPTER 3, Scientific

Measurement(continued) Label each of the three following sentences that describes accuracy with an A.Label each sentence that describes precision with a

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P. _____ 5. Four of five repetitions of a measurement were numerically identical, and the fifth varied from the others in value by less than 1%.

SECTION 3.1 MEASUREMENTS AND THEIR UNCERTAINTY

Use your measurements for the aluminum foil (in Table 2) along with the

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true density of aluminum ($\rho_{\text{Al}} = 2.70 \text{ g/cm}^3$) to calculate the foil thickness, in cm. Consider the foil to be a very flat rectangular box, where
Volume of foil = $V = \text{length} \times \text{width} \times \text{height (thickness)}$. Show your work, and report your answer in ...

1: Measurements in the Laboratory

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(Experiment) - Chemistry ...

Chapter 3 Scientific Measurement.

Adapted from notes by Stephen L.

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Importance of Measurement.

OBJECTIVES: zDistinguish between

quantitative and qualitative

measurements. zConvert measurements

to scientific notation. Measurements.

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Qualitative measurements -.

Quantitative measurements -.

Section 3.1 The Importance of Measurement Scientific ...

Chapter 1 - Measurements in Chemistry.

This content can also be downloaded as an printable PDF or an interactive PDF.

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required for full functionality. ... Section 3: Scientific Notation. The study of chemistry can involve numbers that are very large. It can also involve numbers that are very small.

Chapter 1: Measurements in Chemistry - Chemistry

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