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## **Chemistry Matter Change Chapter 16**

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heat. Law of Conservation of Energy. States that in any chemical reaction or physical process, energy can be converted from one form to another, but it is ...

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concentration of a reactant or product

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per unit.... Atoms, ions, and molecules must collide in order to react Transition state, a temporary, unstable arrangement of atoms i.... The change in concentration of a reactant or product per unit....

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A spontaneous change may be so rapid that it is essentially instantaneous or so slow that it cannot be observed over any practical period of time. To illustrate this concept, consider the decay of radioactive isotopes, a topic more thoroughly treated in the chapter on nuclear chemistry.

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### 16.1 Spontaneity - Chemistry

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31 number of moles of known substance  
C H C H C H O O O CO CO CO  
H O H O H O number of moles of unknown substance  
O CO H O C H CO H O C H O H O C H O CO use mole ratio  
5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...

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16. Which collision(s) formed products? What were the products? B 17. Explain why the other collision(s) did not form products. 18. Which collision(s) formed an activated complex? Identify the activated complex. 98 Chemistry: Matter and Change Chapter 17 Study Guide for Content Mastery

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